

## PRODUCT: SODIUM HYPOCHLORITE (HYPO) REVISION: 7 DATED: 04/09/22 PAGE 1 OF 9

PRODUCT SPECIFICATION				
Product Name Alternative Name Specification Reference	Sodium Hypochlorite Solution, 14-15% Available Chlorine Bleach Liquor HYPO/6 (19/09/IHJK)			
	Units	Specification Range	Typical Analysis	
Available Chlorine	% w/w	>14.0	14.8	
Density at 20°C	kg/litre	1.24 - 1.27	1.25	
Sodium Hypochlorite	$^{0}\!/_{\!0}~{ m W}/{ m W}$	>14.7	15.5	
Sodium Hydroxide	% w/w	0.2 - 1.0	0.7	
Sodium Carbonate	% w/w	<1.5	0.3	
Iron, Fe	ppm	<2	0.4	
Methods of Analysis				
Methods of analysis are given	in BS4426:1969			
Additional Information				

### Additional Information

### **Quality Standard**

This product meets the requirements of BS EN 901:2013 chemicals for the treatment of water intended for human consumption

The strength specification is that at the time of supply. Sodium Hypochlorite solution is inherently unstable and slowly loses strength, producing Sodium Chloride and Oxygen, therefore time/light/foreign matter all contribute (sometimes to significant effect) to a reduction in strength. We recommend no more than 2-3 months stock on hand in cool/dark storage conditions.

## NOTES

## **Exclusion of Liability**

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### **Health and Safety**

A Material Safety Data Sheet has been issued describing the health, safety and environmental properties of this product, identifying the potential hazards and giving advice on the handling precautions and emergency procedures. This must be consulted fully before handling, storage and use.



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## SAFETY DATA SHEET

# 1. IDENTIFICATION OF THE SUBSTANCE/PREPARATION AND COMPANY

1.1 Product Identifier

Trade Name Sodium Hypochlorite

Substance Name Hypo. Bleach, concentrated. Sodium hypochlorite liquor, concentrated.

 Index Number
 017-011-00-1

 CAS Number
 7681-52-9

 EINECS Number
 231-668-3

REACH Registration Number 01-2119488154-34-XXXX

Relevant identified uses of the substance or mixture and uses advised against

Identified use(s) Bleaching agent, oxidising agents, intermediate, Industrial and professional

cleaning

Uses advised against None identified

1.3 Details of the supplier of the safety data sheet

Fluid Science Limited

Unit 3b Arbour Ct, Arbour Lane,

**Knowsley Industrial Park** 

Kirkby L33 7XE

+44 (0) 1244 506 860 sales@fluidscienceltd.com

1.4 Emergency telephone number

Tel: 0870 190 6777 (National Chemical Emergency Centre) +44 (0)1270 502891

## 2. HAZARDS IDENTIFICATION

### 2.1 Classification of the substance or mixture

Aquatic Acute 1: Very toxic to aquatic life.

Aquatic Chronic 2: Toxic to aquatic life with long lasting effects.

Eye Dam. 1: Causes serious eye damage. Met. Corr. 1: May be corrosive to metals.

Skin Corr. 1B: Causes severe skin burns and eye damage.

For the full text of the H-Statements mentioned in this Section, see Section 16

### 2.2 Label elements

### 2.2.1 According to Regulation (EC) No. 1272/2008 (CLP).

Hazard Pictogram





Signal word(s): Danger. Hazard statement(s)

H290: May be corrosive to metals.

H314: Causes severe skin burns and eye damage.

H400: Very toxic to aquatic life.

H411: Toxic to aquatic life with long lasting effects.

## Precautionary statement(s)

P260: Do not breathe vapour

P273: Avoid release to the environment

P280: Wear protective gloves/protective clothing/eye protection/face protection

P301 + P330 + P331: IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

P303 + P361 + P353: IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water/shower.

P305 + P351 + P338: IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

### Additional Labelling:

EUHO31: Contact with acids liberates toxic gas



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### 2.3 Other hazards

None known

## 3. COMPOSITION/INFORMATION ON INGREDIENTS

Substances

- Chemical nature: Sodium hypochlorite Aqueous solution

requeous solution			
Chemical Name	Identification Number		Amount (%)
Sodium hypochlorite, solution	Index No. CAS No. EC No. REACH Reg. No. Classification	017-011-00-1 7681-52-9 231-668-3 01-2119488154-34-xxxx Met. Corr. 1 H290 Skin Corr. 1B H314 Aquatic Acute 1 H400 Aquatic Chronic 2 H411 EUH031	12 – 16%

For the full text of the H-Statements mentioned in this Section, see Section 16.

## 4. FIRST AID MEASURES

### 4.1 Description of first aid measures

#### Inhalation

Remove patient from exposure, keep warm and at rest. Administer oxygen if necessary. Obtain medical attention.

#### Skin contact

Take off immediately all contaminated clothing. Rinse skin with water/shower. Wash contaminated clothing before reuse. If symptoms develop, obtain medical attention.

#### Eye contact

Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Immediately call a POISON CENTRE/doctor.

#### Ingestion

Do NOT induce vomiting. Provided the patient is conscious, wash out mouth with water and give 200-300 ml (half a pint) of water to drink. Immediately call a POISON CENTRE/doctor.

# 4.2 Most import symptoms and effects, both acute and delayed

Inhalation: May cause breathing difficulty. Cough. Gas (chlorine) produced under fire or acidic conditions is toxic by inhalation.

Skin Contact: Causes burns.

Eye Contact: Risk of serious damage to eyes.

Ingestion: Will cause corrosion of and damage to the upper gastrointestinal tract.

# 4.3 Indication of any immediate medical attention and special treatment needed

In cases of severe exposure, pulmonary oedema may develop. Fluid build up on the lung (pulmonary oedema) may occur up to 48 hours after exposure and could prove fatal. Immediately call a POISON CENTER/doctor. Treat symptomatically.

# 5. FIRE FIGHTING MEASURES

# 5.1 Extinguishing Media

Suitable extinguishing media: Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

Unsuitable extinguishing media: High volume water jet

# 5.2 Special hazards arising from the substance or mixture

Non-combustible. May decompose in a fire, giving off toxic and irritant vapours. (chlorine). Chlorine is an oxidising agent.

## 5.3 Advice for fire-fighters

Fire fighters should wear complete protective clothing including self-contained breathing apparatus. Use water spray or fog to knock down and absorb corrosive fumes. Keep fire exposed containers cool by spraying with water. Dike fire control water for later disposal.

### 6. ACCIDENTAL RELEASE MEASURES

### 6.1 Personal precautions, protective equipment and emergency procedures

Provide adequate ventilation. Do not use metal containers for spilled liquid. Wear appropriate personal protective equipment, avoid direct contact.

### **6.2** Environmental precautions

Avoid release to the environment. Spillages or uncontrolled discharges into watercourses must be alerted to the appropriate regulatory body.

## 6.3 Methods and material for containment and cleaning up

Collect spillage. Small spillages: Wash the spillage area with water. Large spillages: Contain spillages with sand, earth



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or any suitable adsorbent material. Earth may be shovelled to contain spillage and to avoid contamination of sewers and watercourses

### 6.4 Reference to other sections

See also Section 8, 13

## 7. HANDLING AND STORAGE

### 7.1 Precautions for safe handling

Avoid contact with skin and eyes. Avoid inhalation of mists. Wear protective gloves/protective clothing/eye protection/face protection. Ensure adequate ventilation.

## 7.2 Conditions for safe storage, including any incompatibilities

For small quantities - vented containers made from glass or PVC are suitable.

For large quantities - glass reinforced plastic tanks with a PVC lining, rubber lined mild steel or high density polyethylene tanks are suitable. Storage tanks should be completely enclosed except for vents and overflows. Provision should be made to wash tanks clear of sludge, which can build up due to salting out of solids during natural decomposition.

Storage temperature: Ambient. Keep away from heat and direct sunlight.

Storage life: Stable under normal conditions.

Incompatible materials: Do not mix with acid. Avoid contact with other cleaning agents.

## 7.3 Specific end use(s)

No further relevant information

## 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

## 8.1 Control parameters

Sodium Hypochlorite Not Listed (UK HSE EH40).

In case of chlorine emission the occupational exposure limit for chlorine should be observed.

Chlorine: CAS No. 7782-50-0 STEL 0.5 ppm 1.5 mg/m<sup>3</sup>

Region Source

Europe EU Occupational Exposure Limits

United Kingdom

Workplace Exposure Limits (WEL) Remark Notes

### **Sodium Hypochlorite solution**

## Derived No Effect Level (DNEL)/Derived Minimal Effect Level (DMEL)

DNEL / DMEL	Oral	Inhalation	Dermal
Industry - Long Term - Local effects		1.55 mg/m <sup>3</sup>	0.5%
Industry - Long Term - Systemic effects		1.55 mg/m <sup>3</sup>	
Industry - Short term - Local effects		3.1 mg/m <sup>3</sup>	
Industry - Short term - Systemic effects		3.1 mg/m <sup>3</sup>	
Consumer - Long Term - Local effects		1.55 mg/m <sup>3</sup>	
Consumer - Long Term - Systemic effects		1.55 mg/m <sup>3</sup>	
Consumer - Short term - Local effects	0.26 mg/kg bw/day	$3.1 \text{ mg/m}^3$	
Consumer - Short term - Systemic effects		3.1 mg/m <sup>3</sup>	

# Predicted No Effect Concentration (PNEC)

Environment	PNEC
	0.21 μg/l Fresh water, 0.042 μg/l Marine water, 0.26 μg/l Intermittent releases, 30 μg/l Sewage treatment plant
Terrestrial Compartment	No data
Atmospheric Compartment	No data

# 8.2 Exposure controls

### Appropriate engineering controls

Provide adequate ventilation, including appropriate local extraction. A washing facility/water for eye and skin cleaning purposes should be present.

## Respiratory protection

Normally no personal respiratory protection is necessary. When required to spray sodium hypochlorite solutions or to work in mists adequate respiratory protection should be provided. Where a cartridge/canister respirator is suitable use: Type B P3



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### Eye protection

Wear eye protection with side protection (EN166). Goggles giving complete protection to eyes. If splashes are likely to occur: Full face shield.

## Skin and body protection

Wear protective clothing and gloves: The following materials are suitable for protective gloves (permeation time >= 8 hours): PVC (0.5mm), Neoprene (0.5mm), Butyl rubber (0.5mm), Nitrile rubber (0.35mm), Natural rubber (0.5mm).

### Thermal hazards

None known.

## **Environmental Exposure Controls**

Spillages or uncontrolled discharges into watercourses must be alerted to the appropriate regulatory body.

## 9. PHYSICAL AND CHEMICAL PROPERTIES

## 9.1 Information on basic physical and chemical properties

Appearance Liquid.

Colour Greenish-yellow
Odour Faintly chlorinous
Odour threshold Not known.

pH >12.5 Melting point/freezing point -17 °C

Initial boiling point and boiling range 110 °C

Flash Point

Evaporation rate

Flammability (solid, gas)

Upper/lower flammability or explosive limits

Vapour pressure

Not applicable.

Not flammable

Not applicable.

2.5 kPa (20°C)

Vapour density 2.5

Density (g/ml) Not available.
Relative density 1.26 approx (20°C)

Solubility(ies) Solubility (Water): Miscible

Solubility (Other) Not available.

Partition coefficient: n-octanol/water Log Pow: -3.42 (20°C)
Auto-ignition temperature Not applicable.
Decomposition Temperature (°C) Not available.
Viscosity 2.6mPa.s (20°C)

Explosive properties 2.0 Not explosive.

Oxidising properties May liberate chlorine under certain conditions: chlorine is an

oxidising agent.

## 9.2 Other information

Molecular weight 74.44 g/mol

## 10. STABILITY AND REACTIVITY

## 10.1 Reactivity

Contact with acids liberates very toxic gas. (chlorine). Chlorine is an oxidising agent

## 10.2 Chemical stability

Stable under normal conditions. Stability of the solution decreases with the action of heat, light and in the presence of some trace impurities.

### 10.3 Possibility of hazardous reactions

Contact with acids liberates very toxic gas. (chlorine). Chlorine is an oxidising agent.

Reacts with ammonia solutions and amines to form explosive compounds. Can react violently if in contact with methanol. Decomposition with evolution of oxygen is accelerated by light and heat and also by contact with many metals, particularly copper, nickel, iron and 'monel'.

# 10.4 Conditions to avoid

Incompatible materials. Keep away from heat and direct sunlight.

# 10.5 Incompatible materials

Decomposition with evolution of oxygen is accelerated by light and heat and also by contact with many metals, particularly copper, nickel, iron and 'monel'.

# 10.6 Hazardous decomposition products

Chlorine. Oxygen.



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## 11. TOXICOLOGICAL INFORMATION

### 11.1 Information on toxicological effects

### **Acute toxicity**

### Ingestion

Not classified.

Data from sodium hypochlorite solution, at the highest industrially produced concentration of around 15%, shows low oral toxicity. LD50 value (rat, oral) used for Chemical Safety Assessment , 1100 mg/kg bw (as available chlorine) Will cause corrosion of and damage to the upper gastrointestinal tract.

#### Skin contact

Not classified.

LD50 (rat) > 20,000 mg/kg bw

#### Inhalation

Not classified.

LC50 (rat) (1 hr) >10,500 mg/m³ (as available chlorine)

## Skin corrosion/irritation

Calculation method: Causes severe skin burns and eye damage.

### Serious eye damage/irritation

Self classification: Causes serious eye damage.

#### Skin sensitisation data

Not classified.

Human Patch testing suggests that sodium hypochlorite is unlikely to be a skin sensitiser. Reliable test data indicates that sodium hypochlorite has no potential for skin sensitisation in animals.

## Respiratory sensitisation data

Not classified.

May be irritant to the respiratory tract.

## Germ cell mutagenicity

Not classified.

On the basis of a weight of evidence approach, sodium hypochlorite should not be classified as genotoxic as the majority of the relevant in-vitro and in-vivo mutagenicity studies were negative.

### Carcinogenicity

Not classified.

On the basis of a weight of evidence approach, sodium hypochlorite has been shown not to be carcinogenic in animal studies or in humans.

### Reproductive toxicity

Not classified.

There is no evidence from animal studies that sodium hypochlorite has any adverse effects on development or fertility.

## Lactation

Not classified.

### **STOT - single exposure**

Not classified. May be irritant to the respiratory tract.

Note: ≥20% solution Classified as irritating to the respiratory system.( STOT SE 3)

### STOT - repeated exposure

Not classified.

Studies in animals have shown that repeated exposures produce no significant effects

# **Aspiration hazard**

Self classification: Not an aspiration hazard

### 12. ECOLOGICAL INFORMATION

# 12.1 Acute Toxicity

## Aquatic invertebrates

### Acute aquatic toxicity

Daphnia magna, Fresh water. EC50 (48 hour): 0.141 mg/l (Crassostrea virginica), Marine water. EC50 (48 hour): 0.026 mg/l Ceriodaphnia dubia, Fresh water. EC50 (48 hour): 0.035 mg/l

## Chronic

Oyster, Marine water. NOEC (7 day): 0.007 mg/l

## Fish

## Acute aquatic toxicity

Fish, Fresh water. LC50 (96 hour): 0.06 mg/l Fish, Marine water. LC50 (96 hour): 0.032 mg/l

### Chronic

Fish, Marine water. NOEC (28 days): 0.04 mg/l

### Algae

### Acute aquatic toxicity

Algae (Pseudokirchnerella subcapitata) (Liedtke, 2013) EC50: 0.04 mg/l Myriophyllum spicatum, Fresh



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water. EC50 (96 hour): 0.1 mg/l

Chronic

Algae (Pseudokirchnerella subcapitata) (Liedtke, 2013)ErC10: 0.03 mg/l; NOEC: 0.017 mg/l

Algae (periphyton), Fresh water. NOEC (7 days): 0.0021 mg/l

**Toxicity** 

Sediment Compartment Not classified.
Terrestrial Compartment Not classified.

### 12.2 Persistence and degradability

Sodium hypochlorite is a strong oxidiser. It will react with organic substances present in soil and sediments and degrades rapidly to chloride. Sodium hypochlorite is substantially removed in biological treatment processes.

## 12.3 Bio accumulative potential

Sodium hypochlorite has low potential for bioaccumulation and decomposes in water. log P (calculated) -3.42

## 12.4 Mobility in soil

Sodium hypochlorite is mobile in soil and sediments.

### 12.5 Results of PBT and vPvB assessment

Not classified as PBT or vPvB.

### 12.6 Other adverse effects

Sodium hypochlorite is substantially removed in biological treatment processes. There is evidence of inhibition to the aerobic treatment process at a concentration (mg/l) of 0.05 mg/l.

# 13. DISPOSAL CONSIDERATIONS

### 13.1 Waste treatment methods

Dispose of contents in accordance with local, state or national legislation. Send to a licensed recycler, reclaimer or incinerator. Dispose of this material and its container to hazardous or special waste collection point.

### Additional information

Disposal should be in accordance with local, state or national legislation.

Disposar should be in decordance with rocal, state of national registation.				
14. TRANSPORT INFORMATION				
14.1 UN Number				
ADR	1791			
RID	1791			
IMDG	1791			
14.2 UN Proper Shipping Name				
ADR	HYPOCHLORITE SOLUTION			
RID	HYPOCHLORITE SOLUTION			
IMDG	HYPOCHLORITE SOLUTION			
14.3 Transport hazard class				
ADR/RID Class	8			
IMDG Class	8			
IMDG EMS	Not available			
ICAO/IATA				
Excepted Quantities	E2			
Passenger and Cargo Aircraft Limited Quantities Packing Instructions	Y840			
Passenger and Cargo Aircraft Limited Quantities Max net Qty	0.5L			
Passenger and Cargo Aircraft Packing Instructions	851			
Passenger and Cargo Aircraft Max net Qty	1L			
Cargo Aircraft Packing Instructions	855			
Cargo Aircraft Max net Qty	30L			
Special Provisions	A3			
Emergency Response Guidebook (ERG) Code	8L			
ADR Classification Code	C9			
ADR HIN	80			
ADR Transport Category	2			
Tunnel Restriction Code	E			
Emergency Action Code	2X			
APP Advice on Additional Personal Protection (APP)	Not applicable			
14.4 Packing group				
Packing group	II			
Labels	8			



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Special provision Limited quantities

Mixed Packing Instructions for Packages Special Packing Provisions for Packages Mixed Packing Instructions for Packages 521 1L P001 IB

MP15

P001 IBC02 PP10 B5

14.5 Environmental hazards

Environmentally hazardous

Classified as a Marine Pollutant

### 14.6 Special precautions for users

Not applicable

## 14.7 Transport in bulk according to Annex II of MARPOL 73/78 and the IBC Code

Product Name SODIUM HYPOCHLORITE SOLUTION

Ship Type 2
Pollution Category Y
Packing Instructions for Portable Tanks T7

Special Provisions for Portable Tanks TP2

TP24 Tank Code for Tanks

Special Provisions for Tanks

Vehicle for Tank Carriage

L4BV(+)

TE11

AT

Special Provisions for Carriage – Packages

Special Provisions for Carriage - Bulk Loading

Special Provisions for Carriage - Loading,

Not applicable

Not applicable

Unloading and Handling

Special Provisions for Carriage – Operation Not applicable

# 15. REGULATORY INFORMATION

# 15.1 Safety, health and environmental regulations/legislation specific for the substance or mixture

European Regulations - Authorisations and/or Restrictions On Use. Not listed

Candidate List of Substances of Very High Concern for Authorisation REACH: ANNEX XIV list of substances subject to authorisation. Not listed

REACH: Annex XVII Restrictions on the manufacture, placing on the market and use of certain

dangerous substances, mixtures and articles. Not listed

Not listed Not listed Not listed

Community Rolling Action Plan (CoRAP) Not listed

Regulation (EC) N° 850/2004 of the European Parliament and of the Council on persistent organic pollutants

Regulation (EC) N° 2037/2000 on substances that deplete the ozone layer. Not listed

Regulation (EU)  $N^{\circ}$  649/2012 of the European Parliament and of the Council concerning the export and import of hazardous chemicals. Not listed

## 15.2 Chemical safety assessment

A Chemical Safety Assessment has been performed

### 16. OTHER INFORMATION

### **Hazard Statements**

H290: May be corrosive to metals.

H314: Causes severe skin burns and eye damage.

H318: Causes serious eye damage.

H400: Very toxic to aquatic life.

H410: Very toxic to aquatic life with long lasting effects.

H411: Toxic to aquatic life with long lasting effects.

## Abbreviations and acronyms

ADN: European Agreement concerning the International Carriage of Dangerous Goods by Inland Waterways

ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road

CAS: Chemical Abstracts Service

CLP: Regulation (EC) No 1272/2008 on classification, labelling and packaging of substances and mixtures

DNEL: Derived No Effect Level EC: European Community

EINECS: European Inventory of Existing Commercial Chemical Substances

IATA: International Air Transport Association

IBC: Intermediate Bulk Container

ICAO: International Civil Aviation Organization



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IMDG: International Maritime Dangerous Goods

LTEL: Long term exposure limit

PBT : Persistent, Bioaccumulative and Toxic PNEC : Predicted No Effect Concentration

REACH : Registration, Evaluation, Authorisation and Restriction of Chemicals RID : Regulations concerning the International Carriage of Dangerous Goods by Rail

STEL: Short term exposure limit STOT: Specific Target Organ Toxicity

UN: United Nations

vPvB: very Persistent and very Bioaccumulative

## Source of key data used to compile the data sheet

Supplier information

# **Modifications from last revision**

The Specification has been updated. The Safety Data Sheet remains the

same. Date 04/09/22